

Principles of Soil Chemistry, Fourth Edition, 2011, 390 pages, Kim H. Tan, 1439813957, 9781439813959, CRC Press, 2011

Learn the secrets of soil chemistry and its role in agriculture and the environment. Examine the fundamental laws of soil chemistry, how they affect dissolution, cation and anion exchange, and other reactions. Explore how water can form water-bridges and hydrogen bonding, the most common forces in adsorption, chelation, and more. Discover how electrical charges develop in soils creating electrochemical potentials forcing ions to move into the plant body through barriers such as root membranes, nourishing crops and plants. You can do all this and more with Principles of Soil Chemistry, Fourth Edition. Since the first edition published in 1982, this resource has made a name for itself as a textbook for upper level undergraduates and as a handy reference for professionals and scientists. This fourth edition reexamines the entire reach of soil chemistry while maintaining the clear, concise style that made previous editions so user-friendly. By completely revising, updating, and incorporating a decade's worth of new information, author Kim Tan has made this edition an entirely new and better book. See what's new in the Fourth Edition

Reexamines atoms as the smallest particle that will enter into chemical reactions by probing new advances testifying the presence of subatomic particles and concepts such as string theory

Underscores oxygen as the key element in soil air and atmosphere for life on earth

Reevaluates the idea of transformation of orthoclase into albite by simple cation exchange reactions as misleading and bending scientific concepts of ion exchange over the limit of truth

Examines the role of fertilizers, sulfur, pyrite, acid rain, and nitrogen fixation in soil acidity, underscoring the controversial effect of nitrification on increasing soil acidity over time

Addresses the old and new approaches to humic acids by comparing the traditional operational concept against the currently proposed supramolecular and pseudomicellar concept

Proposes soil organics, such as nucleic acids of DNA and others, to also adsorb cation ions held as diffusive ion clouds around the polymers

Tan explains, in easy and simple language, the chemical make-up of the four soil constituents, their chemical reactions and interactions in soils as governed by basic chemical laws, and their importance in agriculture, industry, and the environment. He differentiates soil chemistry from geochemistry and physical chemistry. Containing more than 200 equations, 123 figures, and 38 tables, this popular text and resource supplies a comprehensive treatment of soil chemistry that builds a foundation for work in environmental pollution, organic and inorganic soil contamination, and potential ecological health and environmental health risks.

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Soil processes a systematic approach, Sheila Ross, 1989, Technology & Engineering, 444 pages. .

Soil Chemistry, Volume 1 , G.H. Bolt, 1976, Soil chemistry, 281 pages. .

The physical chemistry and mineralogy of soils, Volume 2 , Charles Edmund Marshall, 1964, Science, . .

Humic Matter in Soil and the Environment Principles and Controversies, Kim H. Tan, Apr 22, 2003,

Technology & Engineering, 408 pages. The only book to completely define and explore the genesis, extraction, properties, and impact of humic matter on agriculture, industry, and the environment, Humic Matter in.

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The Chemistry of Soils , Garrison Sposito, Mar 31, 2008, Science, 344 pages. There have been many advances in soil chemistry since Oxford published the first edition of The Chemistry of Soils in 1989. The physical-chemistry approach to soil chemistry.

Soil Chemistry and Its Applications , Malcolm S. Cresser, Ken Killham, Anthony Edwards, 1993, Science, 192 pages. The central role of soil chemistry in the ecosystem and other disciplines is becoming increasingly important. For example the effects of the increased levels of atmospheric.

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Clays , Alain Meunier, Jan 13, 2005, Nature, 472 pages. This monograph presents the fundamentals of the mineralogy and geology of clays. The chemical or isotopic composition and the crystalline structure of clays is based on some.

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